**Green Synthesis, Characterization and Investigating the Antibacterial Potential of Silver Nanoparticles Using Datura (*Datura inoxia*) Against Bovine Mastitis Causing Bacteria**

Saima Qadeer,1\* Asma Ashraf1, Fouzia Bibi1, Muhammad Asad1

***1*** *Department of Zoology, Division of Sciences and Technology, University of Education, Lahore-54000 Pakistan*

\*Corresponding Author:

Dr. Saima Qadeer

[saima.qadeer@ue.edu.pk](mailto:saima.qadeer@ue.edu.pk)

**Abstract:**

**Objective:** The objective of this study was to biosynthesize the AgNPs from the aqueous leaf extract of the *D. inoxia*, characterize them and to evaluate their antibacterial potential against the bacteria associated with bovine mastitis.

**Methods:** The AgNPs were synthesized, purified, optimized and characterized using UV-Vis, XRD, SEM, TEM, HR-TEM and FTIR. Further, In vitro antibacterial activity of the different concentrations of synthesized AgNPs (5, 10, 15, and 20 mg/mL) was assessed against bacterial isolates cultured from milk samples of affected buffaloes, bacteria were identified using standard microbiological techniques.

**Results:** The synthesized AgNPs exhibited a characteristic surface plasmon resonance peak at 330-350 nm, while XRD analysis showed the intense diffraction peaks at 2θ angles of 33°, 46°, and 58°, within the range of 17° and 65°. SEM reveals polydisperse, irregular shaped nanoparticles clusters, further confirmed by TEM as aggregates of nearly spherical to slightly rod-shaped units (20-30nm), consistent with the 24-25 nm crystallite size from XRD and verified by HRTEM. FTIR analysis identify the capping agents that reduce and stabilize the AgNPs. Identified bacterial genera included *Staphylococcus spp*., *Streptococcus spp*., *Enterococcus spp*., *Corynebacterium spp*. (Gram-positive), *E. coli*, *Klebsiella spp*., *P. aeruginosa*, and *Enterobacter spp*. (Gram-negative). The AgNPs exhibited antibacterial activity in the agar well diffusion assay, with the largest zone of inhibition (18.3±0.2mm) observed against *Staphylococcus*at 20 mg/mL of AgNPs. The maximum antibacterial activity of AgNPs was observed at a MIC of 10 µL, particularly against *Staphylococcus* species.

**Conclusion:** These finding suggest that AgNPs synthesized from *D. inoxia* have strongantibacterial activity against mastitis-associated pathogens.

**Keywords:**Bovine Mastitis, AgNPs, *Daura inoxia*, Green Synthesis, nanoparticles

**1. Introduction**

Mastitis in buffalo is a painful inflammation of the udder tissues. It is considered as the highly widespread disease causing economic losses in the dairy sector due to decreased milk production and poor quality 1. Bacteria are primarily responsible for about 70% of cases involved in the etiology of mastitis, the remaining 30% attributed to other agents such as mechanical or physical injury to the gland, etc. 2. Major bacterial agents that cause mastitis include environmental *(Streptococcus agalactiae, E. coli, Enterobacter,* and *Klebsiella)* and infectious *(Staphylococcus aureus* and *Corynebacterium bovis)* 3, 4. The most common method for treating and preventing mastitis is the use of antimicrobial agents, however, misuse has increased the antibiotic resistance 5. The persistence of enterotoxins in milk and milk products, along with the possibility of pathogen transmission, makes it potentially dangerous to human health 6. Due to its complex pathophysiology and emerging resistance, mastitis is sometimes extremely difficult to treat with traditional medications 2.

To reduce the need for antibiotics in mastitis treatment, researchers have explored new therapeutic strategies. The realization that a substance's size could affect its physicochemical properties, highlighted the therapeutic potential of nanoparticles 7. The green synthesized antibacterial nanoparticles have emerged as crucial tools against antibiotic resistant bacteria 8. One of the most widely used nanoparticles is silver nanoparticles (AgNPs) 9, because microorganisms are unable to develop resistance silver, which target multiple sites within the pathogen 10. Because of their large surface area to volume ratio, silver nanoparticles possess enhanced antimicrobial potential 11. AgNPs are widely used to treat microbial diseases because of their strong antiviral, antifungal, and antibacterial properties as well as their antioxidant activity. Intriguingly, most studies report no evidence of toxicity or accumulation in critical organs after topical AgNPs treatment. In order to improve AgNPs' biocompatibility and further reduce their toxicity often associated with chemical reducing agent during preparation used in conventional synthesis, green biosynthesis has recently been adopted 12. AgNPs have numerous uses in the medical sphere, including gene therapy, wounds and burn coverings, dental coatings, and drug transport to specific tissues. In industrial sector, they are used in the production of textiles and cosmetics 13. AgNPs kill bacteria through three ways; (I) leaching of metal ions that directly interact with bacterial cell and cause toxicity 14, (II) stimulate the production of ROS in bacterial cells, oxidative stress results in animal death 15, and (III) disruption of cell membrane permeability that leads to animal death 16. The green synthesis using plant extract has been found to provide additional pharmacological properties to AgNPs 17. Although numerous plant extracts have been used to synthesize the nanomaterials, the field remain active because diverse species can perform as capping and reducing agents, yielding nanoparticles with unique structures and properties 18. Polyphenols and proteins found in plant extract can perform as reducing and capping resources, lowering the valence state of metal ions and giving a precise and stable form to AgNPs 19, 20.

Datura (*D. inoxia*), a shrubby plant from the Solanaceae family, generally recognized as thorn-apple and devil's trumpet, has shown remarkable antioxidant activity21. It is reported that seed extracts from various Datura species effectively scavenge free radicals and neutralize the stable diphenylpicrylhydrazyl (DPPH) 22. It has been documented that this plant is cytotoxic to several cancer cells, and has been utilized in traditional medicine to treat a variety of conditions, such as cutaneous eruptions, colds, and neurological illnesses 23. According to 24, an ethanolic extract of *D. inoxia* shown antiproliferation properties against a range of cancer cells. Specifically extract effectively inhibits the development of the cancer cells by causing death. Research suggested that, *D. inoxia* triggers apoptosis by raising the SubG1 phase and inducing an increase of apoptotic cells. Elevated e markers in LoVo cells further confirm that *D. inoxia* initiate apoptosis. Aqueous leaf extract of *D. inoxia* suppresses K562 cell proliferation in a dose and time-dependent manner, according to the results of cytotoxicity assay. Additionally, normal human B lymphoid cells were less susceptible to *D. inoxia*'s cytotoxicity than K562 cells 25. Furthermore, *D. inoxia* has been documented to have many medical uses, including the treatment of respiratory conditions, soothing antispasmodics, and alleviating pain 26. Additional phytochemicals existing in the plant, like flavonoids, phenols, saponins and glycosides, have been shown to have pain reliever, antibacterial, antipyretic, and anti-inflammatory properties 27, 28. Therefore, it was hypothesized that the biological action of AgNPs can be enhanced by combining them with Datura (*D. inoxia*) leaf extract. The objective of the current study is green synthesis of AgNPs from *D. inoxia*, their characterization and assessment of their antimicrobial action against mastitis causing bacterial.

**2. Materials and Methods**

**2.1. The AgNPs Synthesis:**

The plant species *Datura inoxia,* was identified according to Flora of Pakistan (Taxon Id:109368). Fresh leaves were collected, washed, dried, and ground into a fine powder, and extract was prepared according to 29 and stored at 4°C. Leaf extract was used to reduce AgNO3 solution (1mM) into AgNPs 30. The solution was heated at 70 °C for 1h 29, resulting solution became dark brown, indicating the formation of AgNPs. The approach of 31was used for the purification of AgNPs by diluting them in distilled water and centrifuging (Model Z216MK) them for 5 min at 12,000 rpm.

**2.2. AgNPs Characterization:** A UV visible spectroscopy machine (LX21ODS) used to measure the visual properties of AgNPs within 200–800 nm range. For morphological assessment, AgNPs were examined under a scanning electron microscope (Cube II Emcraft, South Korea). TEM and HRTEM analysis were carried out on JEOL JEM-ARM200F. Metallurgical crystal-like characteristics of AgNPs were evaluated by X-ray diffraction machine (JDX-11, Joel Ltd., Japan) and obtained data were analyzed using software Origin Pro version 9.0. The size of nanoparticles was calculated by applying the Scherrer equation: d= kλ⁄(B Cos θ ), λ= X-ray wavelength, (d=Mean size, k = Dimensionless shape element (Value 0.9), β= Line broadening at half the maximum intensity, θ= Bragg angle). Functional groups were identified by FTIR analysis (Nicolet, Summit LITE, Thermo-Scientific).

**2.3. Bacteria Culturing:**

### Milk sample from Twelve mastitis affected Nili Ravi buffaloes was collected at the Livestock Diagnostic Laboratory, Khushab. The bacteria were separated by serial dilution according to 32, cultured on nutrient agar according to 33. Pure cultures were obtained from mixed cultures using the plate streaking method by 34.

**2.4. Identification of Bacterial Genera:**

Simple stainingwas performed to determine the form and structure of bacteria, using common stains according to method by 35. Gram Staining,which distinguishes bacteria according to the structure of their cell walls was carried out to identify unknown bacteria, given by 36. Motility test was used to investigate the movement of living bacteria according to method given by 37. Catalase test was used to find the catalase enzyme in bacteria according to method given by 38.

**2.5. Antibacterial Analysis:**

Antibacterial activity of nanoparticles was assessed by measuring the values of zones of inhibition (ZOI) using agar wells diffusion 39, 40. Wells of 5 mm were created in the nutrient medium. AgNPs solutions of 5, 10, 15, and 20 mg/mL were made using distilled water, and 25 µL of each was added to wells. Distilled water was used as a control. The mean ZOI and standard deviations were evaluated 41(Leelaprakash & Rose, 2011). The lowest dose of the nanoparticles needed to prevent bacterial growth was determined as the MIC 42.

**2.6. Statistical analysis:**

Data were analyzed using Statistix10. analytical software. Experiment was replicated three time (r=3). The results are shown as mean ± standard deviation. Before analysis, all data were normalized with an arcsine transformation. The results are described as non-transformed means (±SEM). The differences between various groups were calculated by one-way analysis of variance (ANOVA) andf Tukey’s test. A p≤ 0.05 value was accepted as statistically significant.

**3. Results**

**3.1. Green Synthesis of *Datura inoxia* AgNPs**

The change of color from yellowish to brown inside the reaction vessels, specified the synthesis of AgNPs through the reduction of silver. The ideal temperature and pH for AgNPs were assessed to support labor- and time-efficient AgNP production in bulk. pH optimization was performed over a pH range of 2.0 to 9.0 and determined that a pH between 6.0 to 9.0 pH is ideal for AgNPs synthesis, indicating that moderately acidic to slightly alkaline conditions are the most effective. The optimal temperature was found to be between 10° and 100°C. The results showed that the ideal temperature range for green synthesis was between 70° and 90°C.

**3.2. AgNPs Characterization**

The UV-visible spectrophotometry showed taht λmax for *Datura inoxia* AgNPs was recorded between 330-350 nm, with a maximum absorbance of 2.79, as shown in Fig. 1a. The X-ray diffraction (XRD) crystallographic analysis revealed that the AgNPs were polycrystalline, with the XRD spectrum covering 2θ values ranging from 10° to 70°. The prominent peaks for *Datura inoxia* were observed at 33°, 46°, and 58°, corresponding to the (111), (200), and (220) planes, respectively. The XRD results are presented in Fig. 1b. SEM results revealed that the resulting nanoparticles were present in the form of clusters and had irregular shape (Fig. 2 a.). The image depicts polydispersity indicating that they vary in size, which is a common occurrence in green synthesis 43, 44. As the nanoparticles are clusters, the Scherrer equation was used on X-ray diffraction data to determine the mean crystallite size having the size between 24 and 25nm. SEM results were confirmed by TEM Analysis which depicted that these clusters are made up of smaller units that are nearly spherical to slightly rod shaped (Fig. 2 b). Their size range between 20-30nm, that is close to the crystallite size (24-25nm) determined from XRD crystallography. Furthermore, the crystal structure is verified by HRTEM (Fig. 2 c, d). The results of FTIR analysis of this study show different stretches of bonds shown at different peaks (3231, 2913, 1604, 1345, 1313, 1045, 606, and 485 cm−1) (Fig. 1 c).

**3.3. Antibacterial Assay**

Eight bacterial genera were isolated from six milk samples taken from Nili Ravi species of buffalo. These include 1) *Staphylococcus spp*., 2) *Streptococcus spp*., 3) *Escherichia coli*, 4) *Klebsiella spp*., 5) *Pseudomonas aeruginosa*, 6) *Enterococcus spp*., 7) *Enterobacter spp*., and 8) *Corynebacterium spp*. The microorganisms were identified by morphological features and biochemical testing (e.g. simple staining, gram staining, motility test and catalase test). The antibacterial activity of *D. inoxia* AgNPs against the above genera, that cause bovine mastitis, was determined using the diffusion method, as shown in Figure 3. The antibacterial activity was tested at concentrations of 5, 10, 15, and 20 mg/mL of *D. inoxia* AgNPs and control labeled as A, B, C, D, and E respectively on culturing plates. The bacterial inhibition zone was found to be proportional to the concentration of *D. inoxia* AgNPs. As the concentration of *D. inoxia* AgNPs increased, the bactericidal ability also increased, resulting in a larger diameter of the antibacterial ring as shown in Table 1. The study results showed that the maximum zone of inhibition was observed in group D (20 mg/mL of *D. inoxia* AgNPs) for *Staphylococcus spp., Streptococcus spp., Klebsiella spp., Enterococcus spp., Enterobacter spp*., and *Corynebacterium spp*. While, group C (15 mg/mL of *D. inoxia* AgNPs) exhibited the highest zone of inhibition for *Escherichia coli, Klebsiella spp*., and *Pseudomonas spp*. Moreover, the maximum antibacterial activity of AgNPs was observed at a minimum inhibitory concentration (MIC) of 10 µL, particularly against *Staphylococcus* species presented in Fig 4.

**Table 1:** Tabular display of antibacterial activity by *D. inoxia* AgNPs as ZOI (mm)

|  |  |  |  |  |  |  |
| --- | --- | --- | --- | --- | --- | --- |
| Sr. | Bacterial Genera | Zone of Inhibition (mm) | | | | |
| 5 mg/mL | 10 mg/mL | 15 mg/mL | 20 mg/mL | Control |
| 1 | ***Staphylococcus spp.*** | 9.1±0.4**c** | 10.2±0.6**c** | 15.1±0.3**b** | 18.3±0.2**a** | 00±00**d** |
| 2 | ***Streptococcus spp.*** | 8.4±0.2**c** | 9.1±0.5**c** | 10.4±0.4**b** | 11.1±0.3**a** | 00±00**d** |
| 3 | ***Escherichia coli*** | 8.3±0.3**c** | 9.7±0.1**c** | 11.4±0.7**a** | 10.2±0.4**b** | 00±00**d** |
| 4 | ***Klebsiella spp.*** | 7.1±0.3**c** | 9.0±0.4**b** | 10.1±0.2**a** | 10.2±0.6**a** | 00±00**d** |
| 5 | ***Pseudomonas spp.*** | 7.9±0.7**c** | 8.1±0.4**c** | 11.8±0.2**a** | 10.4±0.1**b** | 00±00**d** |
| 6 | ***Enterococcus spp.*** | 8.8±0.2**c** | 8.9±0.6**c** | 10.2±1.1**b** | 15.5±0.3**a** | 00±00**d** |
| 7 | ***Enterobacter spp.*** | 8.1±0.4**c** | 11.6±0.8**b** | 10.4±0.2**b** | 14.1±0.2**a** | 00±00**d** |
| 8 | ***Corynebacterium spp.*** | 7.2±0.5**c** | 8.7±0.1**c** | 9.8±0.2**b** | 11.2±0.6**a** | 00±00**d** |

**4. Discussion**

A green synthesis technique was employed to produce AgNPs using Datura inoxia leaf extract, which act as both reducing and stabilizing agent. The phytochemical present in the extract like flavonoids, phenolic acids, alkaloids, glycosides and saponins jointly acts as effective bio reducer and capping agent, facilitating the formation of stable AGNPs during the green synthesis 25, 27. The appearance of a yellowish-brown color in reaction mixture indicated successful nanoparticle synthesis, consistant with observation reported by 45, and further characterized using UV-Vis spectroscopy, XRD, SEM, FTIR, TEM and HR-TEM microscopy analysis. The observed surface plasmon resonance peak between 330-350 nm and crystalline size range of 24-25 are aligned with previous studies 26, 46 where AgNPs were synthesized from *D. stramonium* leaves and *D. inoxia* flowers 47-49. The magnitude, peak wavelength, and spectral width of a nanoparticle depends on its size, shape, and composition. According to previous studies, the UV-visible absorbance spectrum is influenced by the aggregation state of the nanoparticles 50. A variety of phytoconstituents found in the leaf extract may be responsible for the reduction of metal ions 51. The broad peak at 3231 cm⁻¹, indicates the presence of hydroxyl (-OH) groups (most likely from polyphenols, flavonoids, or water molecules) and amine or amide groups from the *D. inoxia* extract 51. The C-H stretching (alkane groups) vibrations found in lipids, proteins, or organic compounds produced from plants are represented by the peak at 2913 cm⁻¹. The peak at 1604 cm⁻¹ indicates the presence of carbonyl (C=O) functional groups from carboxylic acids, flavonoids, or C=C stretching from aromatic chemicals in plant extracts 46. The peaks at 1345 cm⁻¹ and 1313 cm⁻¹—C-N stretching or C-H bending, respectively indicate the presence of amines from proteins or phenolic chemicals. Alcohols, ethers, esters, or polysaccharides may be the cause of the absorption peak at 1045 cm−1. The vibrations at 1045 cm−1 may be the cause of organo-phosphorous 51.

SEM results showed that the nanoparticles were in the form of clusters and had irregular shape, was confirmed by TEM analysis which depicted that these clusters are made up of smaller units (20-30nm) that are nearly spherical to slightly rod shaped are aligned with previous studies on AgNPs from *D. stramonium* leaves extract showed that in SEM imaging, clumping and formation of nano-sized structures were visible, while TEM analysis, showed spherical nanoparticles with a size of less than 30 nm was seen 52. The narrow peaks in the XRD spectrum indicate a higher crystallinity of the synthesized nanoparticles which are confirmed by HR-TEM. This crystalline structure of the synthesized nanoparticles significantly influences toxicity, as a higher degree of crystallinity correlates with more predictable biological activity. For example, the interaction of nanoparticles with bacterial cells becomes more uniform, enhancing the release of metal ions 53.

The synthesized AgNPs showed concentration dependent antibacterial activity against both gram positive and gram negative bacteria. The highest zone of inhibition (18.3±0.2) was recorded at concentration 20 mg/mL, against *Staphylococcus spp.* These findings are consistent with studies that report higher susceptibility of *Staphylococcus aureus* to AgNPs derived from other plant extract 45, 54. *Enterococcus spp., Enterobacter spp*., also showed significant sensitivity, while *Escherichia coli* and *Pseudomonas spp*. Displayed comparatively lower inhibition zones. This variation may relate to cell wall difference of gram positive and gram negative bacteria. Compared with earlier research, the antimicrobial effects observed here align with those using AgNPs synthesized from other medicinal plants such as *Azadirachta indica* 10, *Ocimum tenuiflorum* 55, and *Datura metel* 5, suggesting that *D. inoxia* is a similarly effective reducing agent for AgNPs with comparable antibacterial properties. Unlike some studies that used higher AgNP concentrations (50–200 μg/mL) 56, 57, the present study achieved notable activity at lower concentrations (5–20 mg/mL), with lower MIC of AgNPs between range of 10-25 µL for bacterial strains test.

**5. Conclusion**

Biosynthesized *D. inoxia* silver nanoparticles (AgNPs) exhibit significant antibacterial activity against multiple mastitis causing bacterial genera, particularly gram positive strains. While the results support their potential as an alternative antimicrobial approach, further research including in vivo trials, toxicity assessments, and long-term stability analysis is essential before clinical or field application can be recommended.

**6. Statements and Declarations**

**6.1. Ethical statement :**The study was approved by Ethical Committee of University of Education Lahore, Pakistan, for the use of animals.

**6.2. Data Availability Statement:** The data that support the findings of this study are available from the corresponding author upon reasonable request.

**6.3. Funding Statement:** Nill

**Authors contribution:** S. Qadeer and F. Bibi designed the study, executed the experiment and drafted the manuscript. M. Asad and A. Ashraf provided expertise in practical work, data analysis and drafting of manuscript.

**Conflicts of Interest:** The authors declare no conflicts of interest.

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**Figure caption**

**Fig. 1.** UV (a), XRD (b) micrographs of *D. inoxia* AgNPs, (c) FTIR micrograph of *D. inoxia* AgNPs

**Fig. 2:** Characterization of*D. inoxia* AgNPs (a) SEM, (b) TEM, (c, d) HR-TEM

**Fig. 3.** (ZOI); a) *Staphylococcus spp., b) Streptococcus spp.,* c) *Escherichia coli, d) Klebsiella spp., e) Pseudomonas aeruginosa*, *f) Enterococcus spp., g) Enterobacter spp., and h) Corynebacterium spp.*

**Figure 4.** MIC of *D. anoxia* AgNPs and error bar shows standard error of mean (n=3).